

RESERVOIR FLUID PROPERTIES SYLLABUS			
Course Title	Reservoir Fluid Properties and Lab		
Course Code	PTE 3315C No. of Credits 3 CR (2 Theor, 1 Prac)		3 CR (2 Theor, 1 Prac)
Department	Departments of Engineering	College	College of Engineering
Pre-requisites Course Code	MEE 3310	Co-requisites Course Code	N/A
Course Coordinator(s)	Hiwa Sidiq		
Email	hiwa.sidiq@komar.edu.iq	IP No.	
Other Course Teacher(s)/Tutor(s)	Non		
Class Hours	MON//WED: 10:00 – 11:30 Room: 103		
Contact Hours	SUN: 13:00 - 16:00		
Course Type	Departmental Requirement		
Offer in Academic Year	Spring 2016		

COURSE DESCRIPTION

This course describe naturally occurring hydrocarbon systems found in the reservoirs as the mixtures of organic compounds that exhibits multiphase over wide ranges of pressures and temperatures. The effect of phase behavior during the life time of the reservoir on production and recovery will be explained in detail. The key oil properties, such as bubble point, GOR, FVF, viscosity etc, will be studied and how these properties are calculated.

COURSE OBJECTIVES

The objective of this course is to help student to be familiar with reservoir fluid properties and how PVT and laboratory data are used for tuning EOS and predictive models. Moreover, PVT envelope analysis and experimental work design.

COURSE LEARNING OUTCOMES

After participating in the course, students should be able to:

- 1. Calculate and measure fluid properties such as density, viscosity etc. (ABET B, D and K)
- 2. Apply thermodynamic aspects to understand reservoir fluids behavior. (ABET A and E)
- 3. Construct phase diagrams for different reservoir fluid type (ABET A and E)
- 4. Calculate fluid properties such as density, specific gravity, compressibility etcl using EOS. (ABET A, and E)
- 5. Apply different EOS methods to calculate fluid properties and compare the results (ABET A, B and E)
- 6. Design PVT tests for reservoir fluids. (ABET A and E)
- 7. Apply the five reservoir fluids concept in production and surface facility design. (ABET A and B)
- 8. Generate the required PVT input data for reservoir simulation using industry standard software. (ABET B, D and K)

*Some of the course outcomes are subjected to PVT lab availability.



RELATED PROGRAM OUTCOMES:

А	An ability to apply knowledge of mathematics, science, and engineering	
В	An ability to design and conduct experiments, as well as to analyze and interpret data	
D	An ability to function on multidisciplinary teams	
E	An ability to identify, formulate, and solve engineering problems	
К	An ability to use the techniques, skills, and modern engineering tools necessary for engineering practice.	

Reference: http://www.abet.org/eac-criteria-2014-2015/

GUIDELINES ON GRADING POLICY

Points	Percentage Scores	Grade
A A-	95–100 90-94	4.0 3.7
B+ B B-	87–89 83-86 80-82	3.3 3.0 2.7
C+ C C-	75–79 70-74 65-69	2.3 2.0 1.7
D+ D D-	60–64 55-59 50-54	1.3 1.0 0.7
F	0–49	0
1	Incomplete Course Work	
W	Official Withdrawal	
Passing Grade is 65% o	r above	

COURSE CONTENTS

Course topics include:

- Introduction to Petroleum Reservoir Fluids
- Compositional Analyses
- PVT Experiments
- Equations of State
- Plus fraction characterisation
- Flash and Phase Envelope Calculations
- PVT Simulation
- Regression to Experimental PVT Data
- Compositional Variations with Depth

*Note: Adding more chapters is governed by the time.



Course Teaching and Learning Activities:

Lectures: during week, the theoretical and practical lectures will be presented throughout the semester; the discussion of practical work within lab will be organized and illustrated with activities.

Assignments: after the lectures, the assignment will be explained and given to students. It is expected to be done on weekly bases.

Quizzes: the contents of each lecture will be discussed during class for open question and answer to make sure every student will participate and be active.

Practical Discussion: during practical session the students will combine together as partners and form a group to discuss their class learning and open tutorial on the topics.

In class brainstorming sessions: provide students with enough sources and background knowledge briefly within the topics during class to top up their challenge packs to be more active.

CLASS REQUIREMENT

- A Scientific Calculator
- Notebook

*Note: Students must bring a notebook, a pen, notebook, calculator, and the periodic table to every class

Assessment Tool	Description	Weight	
Lab experiments and reports	Reports on lab activities and tests (ABET B, D and K)	20 %	
Quizzes	The open question and answer during class (ABET A and E)	10 %	
Assignments	Assignments within the updated topics and presentation (ABET D and J)	10 %	
Tests	Two tests before and after the midterms (ABET A and E)	10 %	
Mid-term exam	Paper examination – all topics that were studied are included (ABET A and E)	20 %	
Final Exam	Examination questions-all topics that were studied during the semester are included (ABET A, and E)	30 %	
ESSENTIAL READINGS: (Journals, textbooks, website addresses etc.)			

Textbooks:

The Properties of Petroleum Fluids McCain, W. D., Penn Well Publishing Co., Tulsa, 2nd Edition, 1990. ISBN 878143351

References:

- PVT and Phase Behaviour of Petroleum Reservoir Fluids, Ali Danesh, 3rd Edition 2003. ISBN 0444821961
- Equations of State and PVT Analysis, Tarek Ahmad 2013. ISBN 0127999787



- Phase behavior, Curtis H. Whitson, Michael R. Brulé 2000. ISBN 1555630871
- The Properties of Gas and Liquids, Bruce E. Poling et al, 2001. ISBN 0070116822

COURSE POLICY (including plagiarism, academic honesty, attendance etc)

KUST Academic Policy

http://sar.komar.edu.iq/files/Student%20hand%20Book%202013.pdf

Attendance:

- Students are expected to attend all lectures and must attend all examinations, quizzes, and practical exercises.
- There is no make-up work for students who miss classes without official permission.
- Student must arrange with the faculty to make-up the missed class.
- Students are subject to the regulation and policies mentioned in the KUST Student Handbook.
- KUST guidelines for lateness are as follows: Three occasions of lateness count as one absence. (You can be considered late the first minute of the lecture time).

GUIDELINES FOR SUCCESS

- 1. Be able to work independently and in groups,
- 2. Pay-attention in the classes is the guarantee of success,
- 3. Extend your knowledge beyond the given textbooks in order to master the subject, and
- 4. Try not to miss the classes



Course calendar: Please check the academic calendar for 2015/2016

Week	Beg/End Dates	Topics	Assessment
1	28 Feb – 3 Mar	 An Introduction into Reservoir Fluid Alkanes, Alkenes, and Alkynes Phase Behavior 	
2	6– 10 Mar	 Phase Behavior of Pure substance Phase behavior two components Phase behavior of three and multicomponents 	Quiz 1 (W1-W2)
3	13– 17 Mar	 Types of phase diagram Ternary diagram Use of ternary diagram Reservoir fluid types Reservoir fluid classifications 	Assignment 1 (19 th Mar)
		20-24 March Nowroz Holiday	
4	27– 31 Mar	 Lab-1: Fluid density measurement: Sample Preparation Density measurement of oil and water 	Practical
5	3– 7 Apr	 Equation of State Ideal gas law Density of ideal gas Kinetic theory of gas Mixture of Ideal gases-partial pressure laws 	Test 1 (7 th Apr) (W1-W3&W5)
6	10– 14 Apr	 Specific gas gravity Real gas behavior Real gas behavior Pseudocritical properties 	Quiz 2 (W5-W6)
7	17–21 Apr	 Lab-2: Viscosity measurement: Viscosity measurement of water and oil, Viscosity measurement using falling ball viscometer. 	Practical
22-28 April Midterm Exam			
8	1– 5 May	 Other EOS for real gas EOS at critical point Virial EOS RK and Peng-Robinson EOS Mixing Ruls The Five Reservoir Fluids Gas FVF 	Assignment 2 (3 rd May)
9	8– 12 May	 Coefficient of gas compressibility Cg for ideal gas and real gas The coefficient of gas viscosity Properties of Black Oil Formation volume factor 	Quiz 3 (W8)



		- Oil GOR	
		- Total FVF - Oil viscosity and interfacial tension	
		Reservoir fluids sample collection	
		- Laboratory tests	
10	15 10 14	1. Compositional measurements	Assignment 3
10	15–19 May	2. CCE	(21 th May)
		3. DL	
		4. Separator test	
		Lab-3; Methods of interfacial tension measurements:	
		Capillary rise method	
	22– 26 May	Wilhelmy plate method	
11		Ring method	Practical
11		 Drop weight method 	Quiz 4 (W9)
		Pendant drop method	
		Interfacial tension (IFT) measurement using	
		pendant drop method	
		 Properties of BO- correlations 	
		- Solution Gas-Oil Ratio	et
10	29 May– 2 Jun	- Oil density	Test 2 (31^{st})
12		- Bubble point	May)
		- Gas Liquid Equilibria	(W8-W10)
		- Vapor-liquid-Equilibrium VLE	
		- Dubble point	
		Pure two and multi component investigation	Practical
13	5 – 9 Jun	\sim CCF DL and flash tests	Flactical
		 Oil viscosity and density calculation 	
14	12– 16 Jun	- Gas Liquid Equilibria	
		- Non ideal fluid	
		- Differential vaporization	Quiz 5 (W12)
		- Surface separation	
15	19 – 23 Jun	Review Week	
26 -30 Jun Final Exam (W1-W15)			

*The above timetable may change and depends on lab availability.